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Solved: 1. Below is a phase diagram for N₂. Use this diagram to determine how the volume changes upon melting for 1 mole of N₂. a. Calculate the change in volume in mL that occurs upon melting. 1. Using the upper portion (between T= -210°C to T=-175°C)

Answer: Chemical Thermodynamics Questions

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Chemical thermodynamics - Wikipedia

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2 3 energy J N m kg m power = = = = time s s s charge current = time = A s energy power = = current*time = A s electrical potential time 2 3 energy kg m electrical potential = = current*time A s electrical potential current = resistance 2 23

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